

## ABSTRACT:

### Tuning the Visible-Light-Drive Photocatalytic Properties of Multi-Decorated TiO<sub>2</sub> by Noble Metals Nanoparticles NPs Towards Both Propionic Acid and NO<sub>x</sub> Degradation

G. Cerrato<sup>1,3</sup>, A. Giordana<sup>1,3</sup>, N. Haghshenas<sup>2,3</sup>, E. Falletta<sup>2,3</sup>, C. L. Bianchi<sup>2,3</sup>

<sup>1</sup> Department of Chemistry, University of Turin, Via Pietro Giuria 7, 10125 Turin, Italy.

<sup>2</sup> Department of Chemistry, University of Milan, via C. Golgi 19, 20133, Milano, Italy.

<sup>3</sup> Cons. Interuniv. Naz. INSTM, via Giusti 9, 50121, Florence, Italy

Unpleasant odors are the most disturbing pollutants and the main reason for air quality complaints in urban and industrial areas. Besides, nitrogen oxide (NO<sub>x</sub>) emissions represent one of the most hazardous air pollutants, causing environmental and health problems, contributing to ground-level ozone, global warming, acid rain, and urban smog [1]. Among the many technologies available, photocatalysis carried out under solar or artificial light has been widely applied to address many air, and not only, pollution issues [2]. First discovered by Fujishima and Honda [3], TiO<sub>2</sub> applications in water and air purification under UV irradiation have multiplied so far: to extend its photo-response to the visible region, thus improving TiO<sub>2</sub> photocatalytic performance, surface modification with noble metals nanoparticles (NPs) has been studied as an efficient approach [4]. Nevertheless, noble metals' high prices and resource shortage limit their applications. In the present research we report the employ of multiple noble metals-modified micrometric TiO<sub>2</sub>-based photocatalysts, prepared by a cheap and sustainable approach based on the use of metal-enriched wastewaters (Ag, Au, Pt) and used for the photodegradation of propionic acid (PA) and NO<sub>x</sub> under LED irradiation. Properly tuning the metal decoration step, the photoactivity of the materials was optimized: in particular, 0.1%Pt @Ag/TiO<sub>2</sub> led to 60% PA removal, whereas the strong PA adsorption on the 0.1%Au @Ag/TiO<sub>2</sub> surface caused a partial deactivation. In contrast, 0.1%Au @Ag/TiO<sub>2</sub> showed the highest photoactivity in the NO<sub>x</sub> decomposition (90%) due to the high tolerance of Au to HNO<sub>3</sub> produced on the catalyst surface.

[1] Y. Wong, et al, Appl. Catal. A Gen. 648 118924 (2022).

[2] Š. Nosek, et al, J. Environ. Chem. Eng., 11, 109758 (2023).

[3] A. Fujishima, K. Honda, Nature, 238, 37 (1972).

[4] N. Rozman, et al, Int. J. Hydrogen Energy, 46, 32871 (2021).